and that the Fe-carbon bonds in 3 average 37 kcal/mol or more, then the condition will be met and the overall process forming 3 will be exothermic. The Fe-carbon bonds in 3 certainly average at least 37 kcal/mol. Hence the proposed structures are reasonable from a thermodynamic viewpoint.

It has been reported that  $D(Cr^+-C_6H_6) \simeq 50 \text{ kcal/mol.}^{15}$ If this is indicative of the strength of the Fe<sup>+</sup>-arene interaction in 2 and the analogous triphenylene complex, then formation of those structures as products of the observed reactions would also be exothermic.

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## Sigmatropic Rearrangement of Vinylallenes: A Novel Route to the 1-Hydroxyvitamin D System<sup>1</sup>

Sir:

We wish to describe our initial results on the preparation and sigmatropic rearrangement of vinylallene intermediates for synthesizing the physiologically important 1-hydroxyvitamin D system.<sup>2-4</sup> As the first example of this approach, we describe a synthesis of 3-deoxy-1-hydroxyvitamin  $D_3$ , whose  $1\alpha$  or 1Sepimer<sup>5</sup> 1a is known to be a highly potent analogue of the

Scheme I OSi(CH<sub>3</sub>)<sub>2</sub>t-Bu 1-111 OSi(CH<sub>2</sub>)<sub>2</sub>t-Bu vi-vii 7 (IS, 7R) 6 (IR, 7R) viii ] VIII 8 (IR, 7S) 9 (18,78)

Reagents: I, Ph<sub>3</sub>PI<sub>2</sub>, Et<sub>3</sub>N, CH<sub>3</sub>CN, reflux 3 hrs. (76%); II, NaBH<sub>4</sub>, EtOH, RT, 3 hrs. (88%); iii, t-Bu(CH<sub>3</sub>)<sub>2</sub>SiCl, imidazole, DMF, RT, 3 hrs. (97%);  $i\nu$ , t-BuLi, -78° (2 hrs.), -30° (1 hr.);  $\nu$ ,  $C_3H_7C\equiv C-Cu\cdot 2P(N(CH_3)_2)_3$ , -78°, 1 hr.;  $\nu$ /, 100b, -78°, 5 hrs.;  $\nu ii$ ,  $Bu_4^{\dagger}N^{\dagger}F^{\dagger}$ , THF, RT, 4 hrs. (37% based on IQb);  $\nu iii$ ,  $h\nu$ . hexane, RT, I hr. (33%).

naturally occurring steroid hormone  $1\alpha,25$ -dihydroxyvitamin  $D_3$  (2a).6

Scheme I summarizes the synthesis of the vinylallenes where the key step is the coupling of cuprate 5 with the ethynyl acetate 10b.7 The latter was prepared from Grundmann's ketone 11a<sup>8</sup> (2b, O<sub>3</sub>/CH<sub>3</sub>CH<sub>2</sub>CHO, 75-92%)<sup>9</sup> by treatment with

Scheme II

$$\underbrace{6,8}_{[1,5]-\text{shift}} \underbrace{1b}_{HO} + \underbrace{\begin{bmatrix} 1,7]-\text{shift} \\ \begin{bmatrix} 1,7]-\text{shift} \end{bmatrix}}_{[1,7]-\text{shift}} \underbrace{\begin{bmatrix} 1,7]-\text{shift} \\ \begin{bmatrix} 1,7]-\text{shift} \end{bmatrix}}_{[1,7]-\text{shift}} \underbrace{\begin{bmatrix} 1,7]-\text{shift} \\ \begin{bmatrix} 1,7]-\text{shift} \end{bmatrix}}_{[1,7]-\text{shift}} \underbrace{\begin{bmatrix} 1,7]-\text{shift} \\ \begin{bmatrix} 1,7]-\text{shift} \end{bmatrix}}_{[2,3]-\text{shift}} \underbrace{\begin{bmatrix} 1,7]-\text{shift} \\ \begin{bmatrix} 1,7]-\text{shift} \end{bmatrix}}_{[3,3]} \underbrace{\begin{bmatrix} 1,7]-\text{shift} \end{bmatrix}}_{[3,3]} \underbrace{\begin{bmatrix}$$

LiC=CH (THF, -78 °C, 90%)<sup>10</sup> and then Ac<sub>2</sub>O (C<sub>5</sub>H<sub>5</sub>N,  $\Delta$ , 36 h, 88%). The coupling of  $5^{13}$  and 10b followed by deprotection<sup>14b</sup> afforded a 37% yield of a mixture of 6 and 7,<sup>14</sup> which was chromatographed to give pure 6 (10%, less polar) and 7 (12%, more polar). 11,14 Photolysis of 6 produced a 1:1 mixture of 6 and 8; similarly 7 gave a 1:1 ratio of 7 and 9.15 These  $Z \rightleftharpoons E$  photoisomerizations proceed analogously to that observed for the known vinylallenes 12 and 13.3 Because 7, 9 and 6, 8 are convertible to 1a<sup>5</sup> and 1b, respectively (below), they are assigned the 1S and 1R configurations, respectively. The allene configurations are tentatively assigned on the basis of spectral comparisons (1H NMR, C-18 methyl) with 12 and 13.<sup>3</sup> The C-18 methyl resonance of 6 or 7 ( $\tau$  9.34 for both) is similar to that of the 7R isomer 12 ( $\tau$  9.35); for 8, 9, and 7Sisomer 13, this same resonance appears at  $\tau$  9.29, 9.28, and 9.27, respectively. Thus, the coupling of vinyl cuprate 5 with **10b**, a formal S<sub>N</sub>2' process, occurs in a completely anti fashion. Certain dialkyl cuprate additions are known to exhibit similar anti selectivity,7b but exceptions are known.7

The allene 6 on thermolysis (refluxing isooctane, 8 h,  $N_2$ ) afforded a ~52% yield of 1b as well as ~5% C and ~4% D. By contrast the  $C_1$  epimer, 7, gave only ~11% of the corresponding vitamin  $1a^5$  accompanied by major amounts of A (~41%) and B (~23%). There was essentially a reversal of product proportions when the photoisomers 8 and 9 were heated: 8 gave only ~9% 1b, but ~45% C and ~18% D; 9 produced ~31% 1a, ~4% A, and ~5% B.  $^{11,14}$ 

The four stereoisomers A-D possess similar spectral properties (UV,  $^1$ H NMR) and are tentatively assigned as 14a, 14b, 15b, and 15a, respectively. The assigned intercyclic diene stereochemistry rests on the rational mechanistic pathway proposed in Scheme II. The major isomer (A or 14a and C or 15b) in each case from heating 7 and 8, respectively, exhibited  $\sim$ 7-Hz bandwidths due to the  $C_1$  proton in the NMR, while the minor isomers (B or 14b and D or 15a, respectively) exhibited larger bandwidths of  $\sim$ 23 Hz. The smaller bandwidth implies that the hydroxyl at  $C_1$  and methyl at  $C_{10}$  are trans diaxially related in 14a and 15b; the larger bandwidth (due to trans- $C_{1,2}$ -proton coupling) indicates that methyl and OH are cis in 14b and 15a with the methyl group fixed axially. <sup>16</sup> Since the  $C_1$  configuration is presumably unchanged during the thermolysis, the  $C_{10}$  configurational assignment follows. <sup>17</sup>

Our original synthesis of 1a from cholesterol required 11 steps (0.2% overall yield).<sup>5</sup> The present convergent synthesis of 1a involves 6-7 steps (0.7% overall yield from readily available vitamin  $D_3$ ); the epimer 1b requires 6 steps (3% overall). The new vinylallene scheme, which will likely improve

on further exploratory studies, is made especially attractive by the fact that a host of A-ring fragments should be obtainable from the commonly available 2-methyl-1,3-cycloalkanediones.

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**Supplementary Material Available:** Spectral and analytical data (7 pages). Ordering information is given on any current masthead page.

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- under nitrogen, freezer).

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# Crossed Molecular Beam Study of the Chemionization Reaction: $SbF_5 + RX \rightarrow SbF_5X^- + R^+$

Sir.

We report the title gas-phase chemionization reaction with RX = benzyl chloride, *tert*-butyl chloride (bromide, iodide), isobutyl iodide, isopropyl iodide, allyl bromide, bromoacetyl